

**quantum theory: a very short introduction - emil kirkegaard** - quantum theory: a very short introduction  
John Polkinghorne has produced an excellent piece of work. . . . many authors of  
popular books on modern physics have the regrettable habit of  
**nitrogen molecules really larger than oxygen molecules ...** - the reason that O<sub>2</sub>, despite a larger mw 32, has a smaller diameter than N<sub>2</sub> mw 28, lies in the electronic structure of the molecules. as indicated by quantum mechanical theory of molecules, the electrons of a molecule form a diffuse  
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page 1 of 28 introduction ultraviolet and visible spectrometers have been in general use for the last 35 years and over this period have become the most  
**lesson 3 chemical bonding molecular orbital theory** - 4 the quantum mechanics of H<sub>2</sub><sup>+</sup> to get a better understanding of bonding, it's best to start with the simplest possible molecule, H<sub>2</sub><sup>+</sup>. what forces do we need to consider?  
**concepts of modern physics - pdx** - concepts of modern physics, sixth edition published by mcgraw-hill, a business unit of the mcgraw-hill companies, inc., 1221 avenue of the Americas, New York, NY 10020.  
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to explain the formation of different kinds of molecules by the combination of different atoms and,  
to understand the origin and nature of  
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